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This brief guidebook assists you in mastering the difficult concept of pushing electrons that is vital to your success in Organic Chemistry. With an investment of only 12 to 16 hours of self-study you can have a better understanding of how to write resonance structures and will become comfortable with bond-making and bond-breaking steps in organic mechanisms.

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Covalent Bonding and Shapes of. Chapter (c) C-Cl <sup>+</sup> <sup>-</sup> C-Cl Chlorine is more electronegative than carbon Problem 1.6 Draw Lewis structures, showing all valence electrons, for these molecules.

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