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CHAPTER 5 REVIEW

The Periodic Law

SECTION 1 SHORT

ANSWER Answer the following questions in the space provided. 1.

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periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

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following questions in
the space provided. 1.
_____ In the modern
periodic table,
elements are ordered
(a) according to
decreasing atomic
mass.

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Section Quiz, continued
Class Date 6. Oxygen
has an electron
configuration of $1s^2 2s^2 2p^4$. In this arrange-
ment, how many inner-
shell electrons does o
ÉIL 7. Which of the
following IS a noble
gas? a. carbon b.
oxygen c. sodium neon

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8. The noble-gas notation for the electron configuration of bromine is $1s^2 2s^2 2p^6 3s^2 3p^4 4s^1$.

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